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Rapid increase in ozone-depleting chloroform emissions from China

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9 **FIRST PARAGRAPH**

10 Chloroform (CHCl_3) contributes to the depletion of the stratospheric ozone layer. However,
11 due to its short lifetime and predominantly natural sources, it is not included in the Montreal
12 Protocol that regulates the production and uses of ozone depleting substances. Atmospheric
13 chloroform mole fractions were relatively stable or slowly decreased during 1990-2010.
14 Here, we show that global chloroform mole fractions increased after 2010, based on in situ
15 chloroform measurements at seven stations around the world. We estimate that the global
16 chloroform emissions grew at the rate of $3.5\% \text{ yr}^{-1}$ between 2010 and 2015 based on
17 atmospheric model simulations. We use two regional inverse modelling approaches,
18 combined with observations from East Asia, to show that emissions from eastern China grew
19 by 49 (41–59) Gg between 2010 and 2015, a change that could explain the entire increase in
20 global emissions. We suggest that if chloroform emissions continuously grow at the current
21 rate, the recovery of the stratospheric ozone layer above Antarctica could be delayed by
22 several years.

23 Large and effective reductions in emissions of long-lived ozone-depleting substances
24 (ODSs) have been achieved through the 1987 Montreal Protocol and its amendments,
25 evidenced by the observed decline in the atmospheric abundances of many ODSs¹. Important
26 remaining uncertainties in the timing of ozone layer recovery are due, in part, to the uncertain
27 impact of very short-lived substances (VSLs), such as dichloromethane (CH₂Cl₂) and
28 chloroform (CHCl₃), which are not currently regulated under the Montreal Protocol¹.
29 Historically, due to the relatively short atmospheric lifetimes (typically <6 months) and
30 therefore low atmospheric concentrations, VSLs have been thought to play a minor role in
31 stratospheric ozone depletion. However, substantial levels of VSLs have been detected in
32 the lower stratosphere^{2, 3, 4} and numerical model simulations suggest a significant
33 contribution of VSLs to ozone loss in the stratosphere^{5, 6, 7}. A recent study shows that CH₂Cl₂
34 atmospheric concentrations are increasing rapidly and, assuming the concentrations continue
35 to grow, the projected CH₂Cl₂ concentrations could substantially delay the Antarctic ozone
36 layer recovery by nearly 30 years, based on global chemical transport model simulations⁸.
37 This study presents the CHCl₃ recent growth, its probable cause and the potential future
38 impact on Antarctic ozone layer recovery.

39 Atmospheric emissions of CHCl₃ are from both natural and anthropogenic sources^{9, 10}.
40 Natural sources are dominated by microbial production in the ocean and soil, with minor
41 contributions from volcanic eruptions. Anthropogenic sources are thought to primarily
42 include HCFC-22 (CHClF₂) and fluoropolymer production, water chlorination and paper
43 manufacturing¹⁰. It has generally been believed that atmospheric CHCl₃ primarily originates
44 from natural sources (e.g., 90%⁹; mainly ocean and soil processes), with only a small
45 anthropogenic contribution. However, recent studies have suggested that anthropogenic

emissions may have been dramatically under-estimated and that ~50% of CHCl_3 emissions may be attributable to these sources^{11, 12}. This study also explores the CHCl_3 sources and their contributions to recent global CHCl_3 changes.

Recent growth in global CHCl_3 mole fractions and emissions

Previous studies using Antarctic firn air showed that Southern Hemisphere polar atmospheric mole fractions increased from 3.7 pmol mol^{-1} in 1920 to a peak of 6.5 pmol mol^{-1} in 1990 before decreasing until the end of the record, in 1997¹¹ (see Figure 1a and b). Based on firn air samples from Arctic and Antarctic sites, Northern Hemisphere mole fractions increased from 5.7 pmol mol^{-1} in 1920, peaked at 17 pmol mol^{-1} in 1990 and decreased after that¹² (Figure 1a and b). There are likely calibration differences between these records, but a general picture of increasing concentrations until 1990 and decreasing concentrations after 1990 emerges from both records. *In situ* baseline measurements (observations with pollution events removed using a statistical filtering algorithm¹³) from the Advanced Global Atmospheric Gases Experiment (AGAGE¹³; see station locations in Figure 1a) show that this downward trend continued until around 2010 at remote sampling locations (Figure 1c). At these AGAGE stations, growth rates between 1995 and 2010 varied between $-0.8\% \text{ yr}^{-1}$ (at American Samoa Observatory in the Southern Hemisphere; SMO) and $-0.3\% \text{ yr}^{-1}$ (at Trinidad Head, California, USA, in the Northern Hemisphere; THD), with the trends observed at other stations (Mace Head, Ireland (MHD); Ragged Point, Barbados (RPB); Cape Grim, Tasmania, Australia (CGO)) lying in between these values. However, we find a renewed growth of global CHCl_3 mole fractions between 2010 and 2015. After 2010, baseline CHCl_3 mole fractions grew in both hemispheres at a higher rate than has been

observed before in the *in situ* or firm records (Figure 1c). During 2010–2015, growth rates for the five AGAGE stations between 2010 and 2015 varied between 2.6% yr⁻¹ at CGO and 6.3% yr⁻¹ at THD. The changes in global mean mole fractions (output from model simulations incorporating measurement data from these five stations; see Methods) were - 0.7% yr⁻¹ during 1995–2010 and increased to 3.9% yr⁻¹ during 2010–2015. The growth in CHCl₃ mole fractions suggests an increase in global CHCl₃ emissions, and the higher rate of increase in the Northern Hemisphere compared to the Southern Hemisphere suggests that the increase in CHCl₃ emissions occurred mainly in the Northern Hemisphere. The abundance of CHCl₃ in the Northern Hemisphere is around 3 times greater than in the Southern Hemisphere, reflecting that the major sources of CHCl₃ are in the Northern Hemisphere.

An atmospheric model has been used to estimate global emissions using CHCl₃ mole fraction data from five non-Asian AGAGE stations (see Methods). The global inversion carried out using this model shows that annual global-total CHCl₃ emissions approximately stabilized at ~271 Gg yr⁻¹, during 2000–2010, with a suggestion of a small decline, and then an increase after 2010, reaching 324 (261–397) Gg yr⁻¹ (16–84 percentiles range) in 2015 (Figure 1c). The average rate of increase in global emissions between 2010 and 2015 was approximately 3.5% yr⁻¹.

Whilst the baseline observations at the five non-Asian AGAGE stations grew between 2010 and 2015, the magnitude of pollution events (i.e. the enhanced mole fractions due to the transport of CHCl₃ from nearby emissions sources, with a magnitude defined here as the 90th percentile of measurements in a year minus the 10th percentile) did not grow significantly at these stations over 2007–2015 (Figure S1 and Table S1). This finding indicates that regional CHCl₃ emissions in Australia (CGO station), the west coast of North America (THD station)

and Europe (MHD station) likely did not increase in this period. In contrast, measurements from Hateruma, Japan (HAT) and Gosan, South Korea (GSN) (see measurement information in Methods for details), show increasing magnitudes of above-baseline pollution events over 2010–2015 (Figure 2a), which suggests an increase in CHCl_3 emissions in eastern Asia, assuming annual mean meteorological conditions have not changed significantly during 2007–2015. Because the observations qualitatively indicate increasing emissions only from eastern Asia, we have focused our analysis on this region in the following sections.

Rapid increase in China's CHCl_3 emissions

To quantify CHCl_3 emissions from eastern Asia, two three-dimensional atmospheric dispersion models (FLEXible PARTicle dispersion model (FLEXPART)¹⁴ and the UK Met Office's Numerical Atmospheric-dispersion Modelling Environment (NAME)¹⁵) were used to simulate the transport of CHCl_3 from potential sources to the measurement locations. Two different inverse modelling approaches were used with each of these models: a Bayesian inversion was used with FLEXPART, and a hierarchical Bayesian “trans-dimensional” approach with NAME¹⁶. For convenience, we label the inversions “FLEXPART” and “NAME” after the transport models, but it should be noted that the statistical approach to inferring fluxes is different with each model (see all details in Methods). Observations and simulations of CHCl_3 mole fractions at the two East Asian stations from FLEXPART and NAME inversions are shown in Figure S2.

Results from both the FLEXPART and NAME inversions show a rapid increase of CHCl_3 emissions from eastern China after around 2010 (Figure 2c). Total CHCl_3 emissions from eastern China were stable during 2008–2010, being 38 (33–44) Gg yr⁻¹ on average.

After that the emissions increased by more than a factor of 2, reaching 88 (80–95) Gg yr⁻¹ (FLEXPART inversion) and 82 (70–101) (NAME inversion) in 2015. CHCl₃ emissions in other East Asian countries/regions were not found to have changed substantially since 2007 (Table S2) and were overall much smaller than eastern China's CHCl₃ emissions. Japan and South Korea rank second and third in this region, with emissions of around 4.7 Gg yr⁻¹ and 1.7 Gg yr⁻¹ on average, respectively. Emissions of CHCl₃ in North Korea and Taiwan in most years were smaller than 1.0 Gg yr⁻¹. Thus, CHCl₃ emissions from eastern China contributed ~87% of East Asian total emissions between 2007 and 2015.

Compared to 2010, 2015 global CHCl₃ emissions increased by 46 (30–61) Gg, while emissions from eastern China increased by 48 (42–54) Gg (FLEXPART inversion) and 50 (41–63) Gg (NAME inversion) (Figure 2d). Therefore, eastern China's emission increase is almost equal to the inferred global emission increase. As mentioned above, CHCl₃ emissions from Australia, North America and Europe likely did not change substantially during this period. Thus, the post-2010 global CHCl₃ emission increase found in this study is most likely due to the rapid emission increase in eastern China, assuming that there are no substantial CHCl₃ emission changes in other regions of the world not covered by the AGAGE measurement network.

Between 2007 and 2015, the highest emissions were inferred for the eastern parts of China, which are highly populated and industrialized. The inferred emissions distribution is broadly consistent with the locations (see Table S3) of factories producing CHCl₃ (Figure 2b from the FLEXPART inversion and Figure S3 from the NAME inversion). However, the exact process or processes responsible for the emissions cannot be identified in this analysis. Between 2010 and 2015, the inferred spatial distribution of CHCl₃ emissions did not change

substantially as emissions rose (see Figure S4). These considerations lead us to conclude that it is most likely that anthropogenic sources are responsible for the rapid emission increase in eastern China during 2010–2015. Recent studies have shown that emissions from eastern Asia of several other ODSs have not declined as expected, or have also increased (e.g., CFC-11¹⁷, CCl₄^{18, 19}, CFC-114 and CFC-115²⁰). It is unclear to what extent, if any, the rise in CHCl₃ emissions from China is related to these findings.

Implications for ozone layer recovery

The Antarctic ozone ‘hole’, a seasonal thinning of the ozone layer above Antarctica during spring, has been predicted to return to pre-1980 levels by around 2050 (± 5 years)¹⁷ or, in more recent studies, towards the end of the century^{18, 19}. These ‘return date’ studies have generally not considered the impact of recent growing levels of the major VSLS, CH₂Cl₂ and CHCl₃. However, a recent study has shown that continued growth of CH₂Cl₂ at current rates could delay Antarctic ozone recovery by several decades⁸. Here, we use the results of this CH₂Cl₂ study to approximate the potential future impact of the increased CHCl₃ on ozone recovery (see Methods). This method relies on the similarities of the CH₂Cl₂ and CHCl₃ lifetime (both 0.4 years²¹) and their atmospheric distribution. We estimate that the increase in CHCl₃ since 2010 could delay Antarctic ozone recovery by ~0.4 year if there is no further growth in CHCl₃ abundance beyond 2015. If growth continued at the average rate observed between 2010 and 2015, the delay could be 4–8 years. If the total increase since 1920 is considered, rather than the increase since 2010, these calculated delays are 1 year and 5–9 years, respectively. Thus, CHCl₃ could have an important role in future ozone layer recovery, especially if mole fractions continue to increase as they have between 2010 and 2015.

Recent studies show that the Asian summer/winter monsoon and the summertime typhoons could provide efficient pathways for directly transporting air pollutants in Asia to the upper troposphere and lower stratosphere^{e.g., 22, 23, 24, 25, 26, 27}. During the late boreal summer and fall, nearly one-fifth of the air in the tropical lower stratosphere had previous contact with the planetary boundary layer over Asia, while negligible fractions originate from over North America and Europe²⁸. Thus, CHCl₃ emitted from East Asia is likely more important for ozone depletion than CHCl₃ emitted from other regions of the world. Demand for CHCl₃ in China is expected to increase in the main application sector - producing polytetrafluoroethylene - at a growth rate of 7%/yr during 2015–2020²⁹ (see details in SI), which is consistent with our scenario of increasing CHCl₃ emissions and mole fractions, at least in the near-term. Considering the above phenomenon of efficient-transport from Asia to the stratosphere and the substantial amount of CHCl₃ emitted in China, the increasing CHCl₃ emissions from China pose a growing threat to ozone layer recovery.

The spatial distribution of our derived emissions strongly suggests that anthropogenic activities, rather than natural sources, are driving the observed rise in China's CHCl₃ emissions. Chloroform and other VSLs are not controlled by the Montreal Protocol, because, apart from CH₂Cl₂, they were previously thought to be mostly from natural sources and to have a minor impact on stratospheric ozone due to their relatively short atmospheric lifetimes. However, this study reveals growing amounts of CHCl₃ from anthropogenic sources in China, with the potential to delay future ozone layer recovery by around 0.4–8 years, depending on whether CHCl₃ abundances pause at the 2015 level, or continue growing at their current rate.

References for main text

1. WMO *Scientific Assessment of Ozone Depletion: 2014. Global Ozone Research and Monitoring Project — Report No. 55*
http://ozone.unep.org/Assessment_Panels/SAP/Scientific_Assessment_2010/index.shtml
(2014)
2. Navarro, M. A., *et al.* Airborne measurements of organic bromine compounds in the Pacific tropical tropopause layer. *Proc. Natl. Acad. Sci. U.S.A.* **112**, 13789–13793 (2015).
3. Sala, S., *et al.* Deriving an atmospheric budget of total organic bromine using airborne in situ measurements from the western Pacific area during SHIVA. *Atmos. Chem. Phys.* **14**, 6903–6923 (2014).
4. Laube, J. C., *et al.* Contribution of very short-lived organic substances to stratospheric chlorine and bromine in the tropics - a case study. *Atmos. Chem. Phys.* **8**, 7325–7334 (2008).
5. Hossaini, R., *et al.* Efficiency of short-lived halogens at influencing climate through depletion of stratospheric ozone. *Nat. Geosci.* **8**, 186–190 (2015).
6. Sinnhuber, B. M., Meul, S. Simulating the impact of emissions of brominated very short lived substances on past stratospheric ozone trends. *Geophys. Res. Lett.* **42**, 2449–2456 (2015).
7. Salawitch, R. J., *et al.* Sensitivity of ozone to bromine in the lower stratosphere. *Geophys. Res. Lett.* **32**, 811–815 (2005).
8. Hossaini, R., *et al.* The increasing threat to stratospheric ozone from dichloromethane. *Nat. Commun.* **8**, 15962–15970 (2017).
9. McCulloch, A. Chloroform in the environment: occurrence, sources, sinks and effects. *Chemosphere* **50**, 1291–1308 (2003).
10. WMO *Scientific assessment of ozone depletion: 2010. Global Ozone Research and Monitoring Project — Report No. 52*
http://ozone.unep.org/Assessment_Panels/SAP/Scientific_Assessment_2010/index.shtml
(accessed January 1, 2012) (2011)

- 217 11. Trudinger, C. M., *et al.* Atmospheric histories of halocarbons from analysis of Antarctic firn
218 air: Methyl bromide, methyl chloride, chloroform, and dichloromethane. *J. Geophys. Res.*
219 *Atmos.* **109**, 310-324 (2004).
- 220
- 221 12. Worton, D. R., *et al.* 20th century trends and budget implications of chloroform and related
222 tri- and dihalomethanes inferred from firn air. *Atmos. Chem. Phys.* **6**, 2847-2863 (2006).
- 223
- 224 13. Prinn, R. G., *et al.* History of chemically and radiatively important atmospheric gases from
225 the Advanced Global Atmospheric Gases Experiment (AGAGE). *Earth Syst. Sci. Data* **10**, 985-
226 1018 (2018).
- 227
- 228 14. Stohl, A., Hittenberger, M., Wotawa, G. Validation of the Lagrangian particle dispersion
229 model FLEXPART against large-scale tracer experiment data. *Atmos. Environ.* **32**, 4245-4264
230 (1998).
- 231
- 232 15. Jones, A., Thomson, D., Hort, M., Devenish, B. The U.K. Met Office's Next-Generation
233 Atmospheric Dispersion Model, NAME III. In: Borrego C, Norman A-L (eds). *Air Pollution*
234 *Modeling and Its Application XVII*. Springer US, 2007, pp 580-589.
- 235
- 236 16. Ganesan, A. L., *et al.* Characterization of uncertainties in atmospheric trace gas inversions
237 using hierarchical Bayesian methods. *Atmos. Chem. Phys.* **14**, 3855-3864 (2014).
- 238
- 239 17. Montzka, S. A., *et al.* An unexpected and persistent increase in global emissions of ozone-
240 depleting CFC-11. *Nature* **557**, 413-417 (2018).
- 241
- 242 18. Lunt, M. F., *et al.* Continued Emissions of the Ozone-Depleting Substance Carbon
243 Tetrachloride From Eastern Asia. *Geophys. Res. Lett.* **0**, (2018).
- 244
- 245 19. Park, S., *et al.* Toward resolving the budget discrepancy of ozone-depleting carbon
246 tetrachloride (CCl₄): an analysis of top-down emissions from China. *Atmos. Chem. Phys.* **18**,
247 11729-11738 (2018).
- 248
- 249 20. Vollmer, M. K., *et al.* Atmospheric histories and emissions of chlorofluorocarbons CFC-13
250 (CClF₃), ΣCFC-114 (C₂Cl₂F₄), and CFC-115 (C₂ClF₅). *Atmos. Chem. Phys.* **18**, 979-1002
251 (2018).
- 252
- 253 21. WMO *Assessment for Decision-Makers: Scientific Assessment of Ozone Depletion: 2014.*
254 *Global Ozone Research and Monitoring Project—Report No. 56* (2014)
- 255

- 256 22. Hossaini, R., *et al.* A multi-model intercomparison of halogenated very short-lived
257 substances (TransCom-VSLS): linking oceanic emissions and tropospheric transport for a
258 reconciled estimate of the stratospheric source gas injection of bromine. *Atmos. Chem.*
259 *Phys.* **16**, 9163-9187 (2016).
- 260
261 23. Yu, P., *et al.* Efficient transport of tropospheric aerosol into the stratosphere via the Asian
262 summer monsoon anticyclone. *Proc. Natl. Acad. Sci. U.S.A.* **114**, 6972-6977 (2017).
- 263
264 24. Randel, W. J., *et al.* Asian Monsoon Transport of Pollution to the Stratosphere. *Science* **328**,
265 611-613 (2010).
- 266
267 25. Vogel, B., *et al.* Fast transport from Southeast Asia boundary layer sources to northern
268 Europe: rapid uplift in typhoons and eastward eddy shedding of the Asian monsoon
269 anticyclone. *Atmos. Chem. Phys.* **14**, 12745-12762 (2014).
- 270
271 26. Oram, D. E., *et al.* A growing threat to the ozone layer from short-lived anthropogenic
272 chlorocarbons. *Atmos. Chem. Phys.* **17**, 11929-11941 (2017).
- 273
274 27. Ashfold, M. J., *et al.* Rapid transport of East Asian pollution to the deep tropics. *Atmos.*
275 *Chem. Phys.* **15**, 3565-3573 (2015).
- 276
277 28. Orbe, C., Waugh, D. W., Newman, P. A. Air-mass origin in the tropical lower stratosphere:
278 The influence of Asian boundary layer air. *Geophys. Res. Lett.* **42**, 4240-4248 (2015).
- 279
280 29. Qianzhan. China's fluoropolymer production capacity will be 230,000 tons by 2020, and PTFE
281 will account for 70% (in Chinese). 2017. Available from:
282 <https://www.qianzhan.com/analyst/detail/220/170629-c33a2ca7.html>
- 283

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Author contributions

X.F. and R.G.P. were responsible for the overall project design. M.R. contributed to the global emission estimation. X.F. contributed to the FLEXPART-based inversions. A.L.G., R.T., A.M., and M.L. contributed to NAME-based inversions. T.S. and Y.Y. provided the CHCl₃ measurement data from HAT, Japan. S.P. and S.L. provided the CHCl₃ measurement data from GSN, South Korea. Other co-authors provided the CHCl₃ measurement data from the five AGAGE stations and the SIO/NIES calibration scales intercomparison data. The manuscript was written by X.F., M.R., A.L.G., J.M., and P.J.F. with contributions from all co-authors.

Competing financial interests

The authors declare no competing financial interests.

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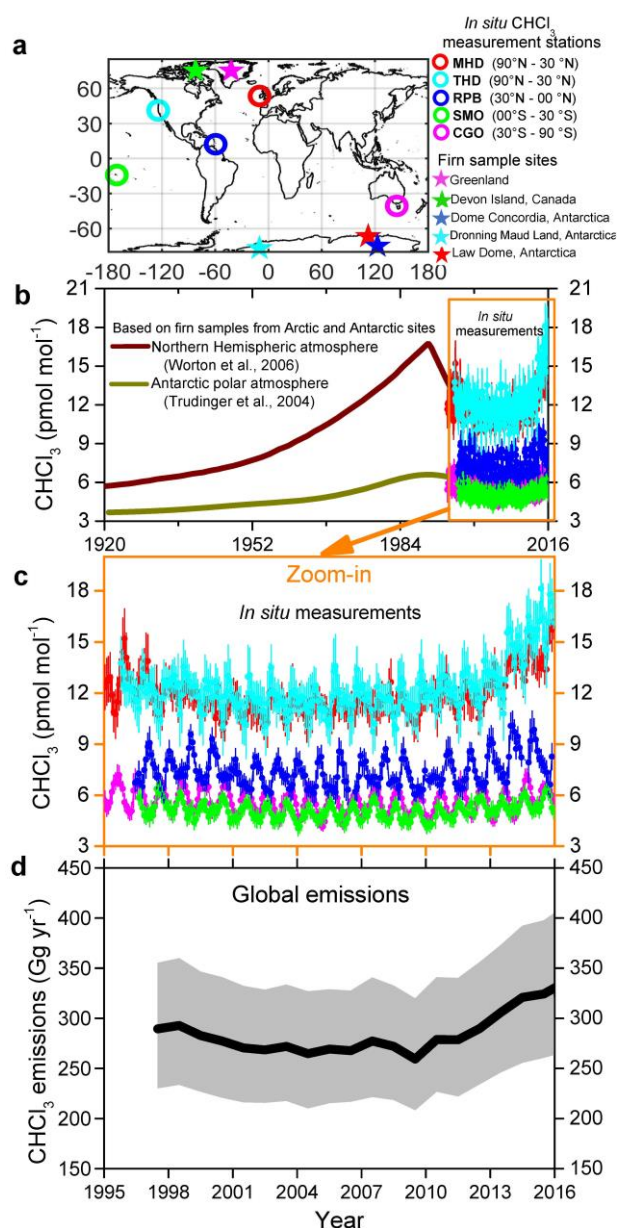
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Additional information

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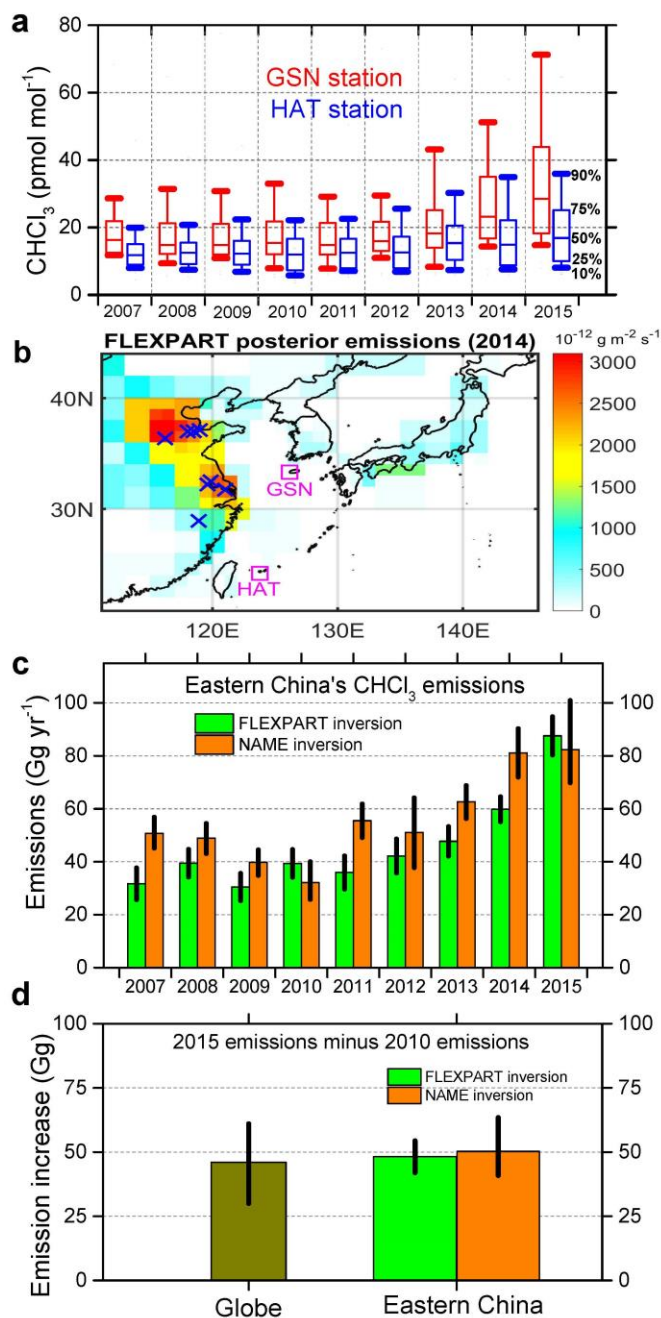
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325

326 **Figure 1. Global atmospheric CHCl_3 mole fractions and emissions.** **a**, Locations of AGAGE
 327 CHCl_3 measurement stations used in the global study. **b**, Northern Hemispheric CHCl_3 mole
 328 fractions, based on CHCl_3 mole fractions measured in Arctic and Antarctic firn air¹², and Southern
 329 Hemispheric polar CHCl_3 mole fractions, based on CHCl_3 mole fractions measured in Antarctic firn
 330 air¹¹; measured CHCl_3 mole fractions (baseline data) at the AGAGE stations. **c**, Measured CHCl_3
 331 mole fractions at the five remote AGAGE stations during 1995–2015. **d**, Global CHCl_3 emissions
 332 (black line) and their uncertainties (gray shaded area; 16–84 percentiles) derived from global inverse
 333 modeling.



334

335 **Figure 2. Measured atmospheric CHCl_3 mole fractions at Hateruma (HAT) and Gosan (GSN)**
 336 **stations and estimated emissions in eastern China from regional inverse modeling. a, Box and**
 337 **whisker plots of atmospheric CHCl_3 mole fractions measured at HAT and GSN stations. b, Map of**
 338 **posterior CHCl_3 emissions derived from regional inverse modeling using FLEXPART model. The**
 339 **blue crosses represent the factories of CHCl_3 production. c, Eastern China's yearly total CHCl_3**
 340 **emissions derived from FLEXPART (1σ uncertainty) and NAME (5–95 percentile range uncertainty)**
 341 **regional inverse modeling. d, 2015 emissions minus 2010 emissions for the total globe and for**
 342 **eastern China.**

Methods

CHCl₃ measurements and global emissions estimations. Atmospheric mole fractions of CHCl₃ are measured at five remote non-Asian AGAGE stations (see information on the five stations in Table S4), using gas chromatography with electron capture detection (GC-ECD) analytical techniques¹³. Global emissions were estimated using baseline mole fractions at the five AGAGE stations and an atmospheric box model^{30, 31}. The model separates the atmosphere into four equal-mass zonal bands (90°N-30°N-0°-30°S-90°S), with vertical divisions at 500 hPa and 200 hPa. The inversion used a Bayesian framework in which the rate of change of emissions was constrained by a prior estimate³². The prior emissions growth rate was assumed to be zero, with an uncertainty in the emissions growth of $\pm 20\%$ of the global total emissions, based on Xiao et al.³³. Uncertainties in the derived fluxes include those due to the observations, the prior constraint, and the atmospheric lifetime (following Rigby et al.³⁴). More information is provided in Supplementary Information.

CHCl₃ measurements and emissions estimation for East Asia. Atmospheric mole fractions of CHCl₃ measured at two Asian stations (HAT and GSN, see Figure 2b) were used in an inverse modeling study to derive East Asian emissions. The HAT station (24.1°N, 123.8°E) is located on a small island at the southern edge of the Japanese archipelago and to the east of Taiwan. CHCl₃ mole fractions in air are measured using a technique based on cryogenic pre-concentration and a capillary chromatograph–mass spectrometry (GC-MS)^{35, 36}. The GSN station (33.3°N, 126.2°E) is situated on Jeju Island south of the Korean Peninsula, and mole fractions of CHCl₃ are measured using the Medusa GC-MS technology³⁷. The time resolution of CHCl₃ measurements is every hour at HAT and every two hours at GSN. The HAT CHCl₃ measurements are reported

in NIES-11 calibration scale and in the SIO-98 calibration scale for GSN data. HAT CHCl₃ data were converted to the SIO-98 calibration scale using the NIES-11/SIO-98 ratio of 1.066+/-0.005.

Two inverse modeling techniques were used to resolve the regional emissions. One is a FLEXPART-based Bayesian inversion. Backward simulations from the FLEXPART model^{14, 38} were driven by meteorological data (European Centre for Medium-Range Weather Forecasts - ECMWF). The backward simulations established a source–receptor relationship matrix, hereafter called “emission sensitivities”. For computational efficiency, we assumed that source-receptor relationships for CHCl₃ were the same as for an unreactive gas. Our tests show that inferred East Asian total CHCl₃ emissions would change by only 1% if model runs that included CHCl₃ reactions with the OH radical were used. The FLEXPART model sensitivities were combined with a Bayesian optimization technique to derive the emission strengths in grid cells in East Asia. The cost function to be minimized is

$$J(x) = \frac{1}{2}(x - x_a)^T S_a^{-1}(x - x_a) + \frac{1}{2}(y^{obs} - Hx_a)^T S_o^{-1}(y^{obs} - Hx_a).$$

We find this minimum by solving $\nabla_x J(x) = 0$, which yields

$$x = x_a + S_a H^T (H S_a H^T + S_o)^{-1} (y^{obs} - Hx_a), \text{ and}$$

$$S_b = (H^T S_o^{-1} H + S_a^{-1})^{-1}.$$

Here x is the state vector of emission strength (g/m²/s) in each grid cell, y^{obs} is CHCl₃ measurement vector, x_a is the prior emission vector, H is the emission sensitivity matrix derived from the FLEXPART backward simulation, S_a is the prior emission error covariance matrix, S_b is the posterior emission error covariance matrix, and S_o is the observational error covariance matrix. We set a uniform prior emission (x_a) distribution over continents and oceans (Figure S5),

so that the posterior emissions are constrained from the measurement data. More information on constructing \mathbf{x}_a , \mathbf{S}_a and \mathbf{S}_o is provided in the Supplementary Information.

The second inversion method employed is a NAME-based hierarchical Bayesian inversion^{16, 39}. NAME is the UK Met Office Lagrangian Particle Dispersion Model (LPDM), which was used here to simulate atmospheric transport. NAME was driven by meteorological information from the Met Office Unified Model, with spanned resolutions of 0.234°–0.563° (longitude) and 0.156°–0.375° (latitude) for 31–70 vertical levels over the period 2007–2015. NAME was run in backward-mode, for a maximum of 30 days backwards in time. For each observation, this resulted in sensitivity maps quantifying the relationship between surface (defined as 0–40 meters above ground level) emissions and concentrations at that receptor. The inversion using the NAME model used a hierarchical Bayesian methodology in which a set of “hyperparameters” comprising model-measurement uncertainties and prior emissions uncertainties were estimated simultaneously with fluxes. In addition, by employing a reversible jump trans-dimensional Markov chain Monte Carlo (TDMCMC) scheme, the spatial decomposition of the underlying flux field was also allowed to vary (i.e. the inversion grid over which fluxes are estimated), allowing the data to derive the resolution with which fluxes were inferred³⁹. More information on NAME inversions is provided in the Supplementary Information.

Due to the emission sensitivity map coverage (Figure S6), emissions are only summed for a specific country/region within the domain between latitude: 20.8°N–44.0°N, 111.2°E– 146.0°E where our observations had significant sensitivity to potential sources.

Estimation of CHCl₃ impact on the recovery of Antarctic stratospheric ozone. Hossaini et al. constructed several scenarios showing the delay in ozone layer recovery (back to 1980 levels) due to various future elevated CH₂Cl₂ levels, compared to a reference scenario with zero

CH₂Cl₂⁸. They found that, to good approximation, Antarctic column ozone changes linearly with the Cl_y load, and their simulations can be used to approximate the impact on ozone of other chlorinated VSLs of similar lifetime (the lifetime of CH₂Cl₂ and CHCl₃ are very similar, at 0.4 years²¹). Here, we make the further simplification that, for relatively small changes in column ozone, the delay in ozone layer recovery can be scaled with Cl loading. Therefore, we use the Hossaini et al. results to calculate the potential delay in ozone recovery due to CHCl₃ by scaling their derived delay in proportion to increase in Cl loading due to CHCl₃ relative to CH₂Cl₂ in 2050. We explore a “continued growth” scenario that the atmospheric CHCl₃ mole fraction grows at 2010–2015 rates. In 2015, the mole fraction of CHCl₃ was ~15 pmol mol⁻¹ in the Northern Hemisphere, as presented Figure 1b, and would reach ~40 pmol mol⁻¹ in 2050. To investigate the impact of only the recent rise in CHCl₃ emissions, 2010 mole fractions (~12 pmol mol⁻¹ in the Northern Hemisphere as presented in Figure 1b) are subtracted. In Scenario 1 in the Hossaini et al. paper, the mole fraction of CH₂Cl₂ in the Northern Hemisphere was ~60 pmol mol⁻¹ and would reach ~165 pmol mol⁻¹ in 2050⁸. This scenario in Hossaini showed a delay in ozone recovery between 30 years (based on chemical transport model simulations) and 17 years (based on chemistry-climate model simulations). Thus, the Antarctic stratospheric ozone recovery could be delayed due to continuous growth of CHCl₃ by approximately 4 (17x(40-12)/165x(3/2)) and 8 (30x(40-12)/165x(3/2)) years (the factor of 3/2 is because CHCl₃ has three Cl atoms per molecule and CH₂Cl₂ has two). Whilst CHCl₃ demand and production in China is projected to increase in near-term²⁹ (see details in SI), future emissions trends are of course highly uncertain. Therefore, we present an alternative scenario (“constant mole fraction”), which assumes that future CHCl₃ mole fractions will remain at the level of ~15 pmol mol⁻¹, as observed in 2015. The corresponding Scenario 3 (no future growth of CH₂Cl₂ mole fractions of ~60 pmol

mol⁻¹ through 2050) in Hossaini et al. showed a 5-year delay in recovery compared to their baseline scenario. Thus, the ozone layer recovery delay in this “constant 2015 mole fraction” scenario for CHCl₃ is about 0.4 (5x(15-12)/60x3/2 =0.4) years. In addition to the assumptions described above, these calculations are thought to be weakly sensitive to differences in the lifetime and atmospheric distribution of CHCl₃ and CH₂Cl₂. Furthermore, they rely on simulations from only the two models used in Hossaini et al.

Data availability. The data that support the findings of this study are available from the corresponding author upon request. CHCl₃ measurement data for East Asia can be accessed by contacting data leads: S.P. (sparky@knu.ac.kr) for GSN and T.S. (saito.takuya@nies.go.jp) for HAT. CHCl₃ measurement data for the five non-Asian AGAGE stations (CGO, SMO, RPB, THD, MHD) used in this study are available at <http://agage.mit.edu/data/agage-data/>.

Code availability. Code for the AGAGE 2-D atmospheric 12-box model and inversion is available upon request from M.R.. Code for the Lagrangian particle transport model (FLEXPART) is available at <https://www.flexpart.eu/>. Code for the UK Met Office’s Numerical Atmospheric-dispersion Modelling Environment model (NAME) is available at <https://www.metoffice.gov.uk/research/modelling-systems/dispersion-model> or upon request from A.J.M.. Code for the FLEXPART-based Bayesian inversion is available upon request from X.F.. Code for the NAME-based hierarchical Bayesian inversion is available upon request from M.R. and A.L.G..

References for Methods

30. Rigby, M., *et al.* Re-evaluation of the lifetimes of the major CFCs and CH₃CCl₃ using atmospheric trends. *Atmos. Chem. Phys.* **13**, 2691-2702 (2013).

- 454
455 31. Cunnold, D. M., *et al.* The Atmospheric Lifetime Experiment: 3. Lifetime methodology and
456 application to three years of CFC13 data. *J. Geophys. Res. Oceans* **88**, 8379-8400 (1983).
- 457
458 32. Rigby, M., Ganesan, A. L., Prinn, R. G. Deriving emissions time series from sparse atmospheric
459 mole fractions. *J. Geophys. Res. Atmos.* **116**, 306-310 (2011).
- 460
461 33. Xiao, X. Optimal Estimation of the Surface Fluxes of Chloromethanes Using a 3-D Global
462 Atmospheric Chemical Transport Model. Ph.D. thesis, Massachusetts Institute of Technology,
463 2008.
- 464
465 34. Rigby, M., *et al.* Recent and future trends in synthetic greenhouse gas radiative forcing.
466 *Geophys. Res. Lett.* **41**, 2623-2630 (2014).
- 467
468 35. Enomoto, T., Yokouchi, Y., Izumi, K., Inagaki, T. Development of an analytical method for
469 atmospheric halocarbons and its application to airborne observation (in Japanese). *J. Jpn. Soc.*
470 *Atmos. Environ.* **40**, 1-8 (2005).
- 471
472 36. Yokouchi, Y., *et al.* High frequency measurements of HFCs at a remote site in east Asia and their
473 implications for Chinese emissions. *Geophys. Res. Lett.* **33**, 814-817 (2006).
- 474
475 37. Miller, B. R., *et al.* Medusa: A sample preconcentration and GC/MS detector system for in situ
476 measurements of atmospheric trace halocarbons, hydrocarbons, and sulfur compounds. *Anal.*
477 *Chem.* **80**, 1536-1545 (2008).
- 478
479 38. Stohl, A., Forster, C., Frank, A., Seibert, P., Wotawa, G. Technical note: The Lagrangian particle
480 dispersion model FLEXPART version 6.2. *Atmos. Chem. Phys.* **5**, 2461-2474 (2005).
- 481
482 39. Lunt, M. F., Rigby, M., Ganesan, A. L., Manning, A. J. Estimation of trace gas fluxes with
483 objectively determined basis functions using reversible-jump Markov chain Monte Carlo. *Geosci.*
484 *Model Dev.* **9**, 3213-3229 (2016).

Supplementary Information

Rapid increase in ozone-depleting chloroform emissions from China

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Global CHCl₃ emission estimation

Global emissions were calculated using baseline atmospheric data from five remote AGAGE stations (MHD, THD, RPB, SMO, CGO). Baseline monthly means were estimated by statistically filtering the high-frequency data, as described by O'Doherty et al.¹. The data were averaged into semi-hemispheres (30°N–90°N, 0°N–30°N, 30°S–0°S, 90°S–30°S) for comparison with mole fractions predicted by the AGAGE 12-box model, which resolves four semi-hemispheres, with vertical levels separated at 500 and 200 hPa^{2,3}. The model uses annually repeating meteorology and OH concentrations from Spivakovsky et al.⁴, tuned to match the growth rate of methyl chloroform. A temperature dependent rate constant for the reaction of CHCl₃ with tropospheric OH from Burkholder et al.⁵ was used, leading to a lifetime in the model of 0.6 years. A Bayesian framework was used to derive emissions from the data and the model, in which prior estimate of the emissions growth rate was adjusted to bring the model into agreement with the data (following Rigby et al.⁶). For this work, our prior estimate was assumed to be zero emissions growth from one year to the next, with a 1-sigma uncertainty in this assumption somewhat arbitrarily chosen to be 20% of the Xiao et al. bottom-up emissions estimate⁷. The inversion propagates uncertainties in the observations through to the derived fluxes, and augments the derived fluxes with uncertainties due to the lifetime and potential errors in the calibration scale⁸.

FLEXPART-based regional emission inversion

In the FLEXPART-based inversion, a Bayesian inversion technique was used. See the equations in main text. \mathbf{x}_a is the prior emission vector. There is no gridded emission inventory available for

CHCl₃. Thus, we used a spatially and temporally uniform prior distribution across the land and ocean domains. The prior total East Asian CHCl₃ emissions from land were set to the amount estimated by Li et al.⁹ (49.5 Gg yr⁻¹). The prior total emissions from the ocean were set to 168 Gg yr⁻¹, according to estimates by Xiao et al.⁷. The map of prior emissions is shown in Figure S5. Prior emissions were the same for all years during 2007–2015. Variable-resolution emission grid boxes were used in the inversions. Following the allocation method by Stohl et al.¹⁰, grid sizes ranged from 24°×24° to 1°×1°, with fine resolution in regions with high emission sensitivity and emission strength (e.g., Eastern China, South Korea, and Japan) and coarse resolutions in remote regions (e.g., Western China and ocean). \mathbf{S}_a is the prior emission error covariance matrix. There is no knowledge of prior emissions and their uncertainties. Here we set the prior emission uncertainty to be 500% of the emission in each grid box, squared values of which are the diagonal elements of \mathbf{S}_a .

The 2-hourly (GSN) and hourly (HAT) CHCl₃ measurement data were averaged into daily means and then assimilated in the inversions. The observational error covariance matrix \mathbf{S}_o was equal to the sum of $\mathbf{S}_{observation}$ and $\mathbf{S}_{emis_aggregation}$ ^{11, 12, 13}

$$\mathbf{S}_o = \mathbf{S}_{observation} + \mathbf{S}_{emis_aggregation}$$

The diagonal elements of $\mathbf{S}_{observation}$ are squared $\sigma_{observation}$. Calculation of $\sigma_{observation}$ is as follows

$$\sigma_{observation} = \sqrt{\sigma_{obs_precision}^2 + \sigma_{obs_representation}^2 + \sigma_{obs_intercalibration}^2 + \sigma_{background}^2}$$

Here $\sigma_{obs_precision}$ is the precision of CHCl₃ measurement (1% and 1% were used for HAT and GSN stations, respectively), $\sigma_{obs_representation}$ is the uncertainty of how representative the

546 observations are and we used one-sigma standard deviation of the observations that were
 547 averaged for each daily value, $\sigma_{obs_intercalibration}$ is 0.005/1.066=0.47% for HAT station (NIES-
 548 11/SIO-98 ratio of 1.066+/-0.005), $\sigma_{background}$ is the variation of background defined by
 549 different methodologies or setups (the mole fraction background filtering method for the
 550 FLEXPART inversion is the same as described in detail by Stohl et al.¹⁰; here, we used various
 551 setups in its calculation, e.g., 5 days or 8 days window to calculate the $\sigma_{background}$). The
 552 observation errors $\sigma_{observation}$ were assumed to be uncorrelated, since the measurement data
 553 used in FLEXPART inversions were daily averages. Consequently $\mathbf{S}_{observation}$ only has
 554 diagonal elements.

555 $\mathbf{S}_{emis_aggregation}$ is the aggregation error from aggregating the spatial emissions from fine
 556 grid cell resolution to variable resolution grid. Calculation of $\mathbf{S}_{emis_aggregation}$ follows the
 557 method of Kaminski et al.¹¹ and Thompson et al.¹²:

$$558 \quad \mathbf{S}_{emis_aggregation} = \mathbf{H} \mathbf{P} \mathbf{S}_{a_fine} \mathbf{P}^T \mathbf{H}^T$$

559 where \mathbf{S}_{a_fine} is the prior emission error covariance matrix at fine grid cell resolution (while \mathbf{S}_a
 560 is the prior emission error covariance matrix at variable-resolution grid). \mathbf{P} represents the
 561 projection of the loss of information in the variable resolution grid compared to the fine grid,
 562 which can be calculated following the equations in Thompson et al.¹². Since $\mathbf{S}_{emis_aggregation}$
 563 has off-diagonal elements, \mathbf{S}_o (sum of $\mathbf{S}_{observation}$ and $\mathbf{S}_{emis_aggregation}$) also has off-diagonal
 564 elements.

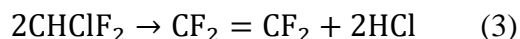
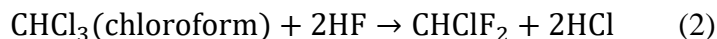
NAME-based regional emission inversion

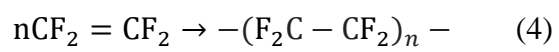
Where possible, the set up for the NAME-based inversion was matched to the FLEXPART inversion method, including use of the same prior emissions field. Here, we describe differences between the NAME and FLEXPART inversion methodologies. In the NAME-TDMCMC inversion, the spatial grid (i.e. the number and placement of resolved regions) over which the flux field was estimated was allowed to vary within a sub-domain (20.8°N–44.0°N, 111.2°E–146.0°E). Surrounding this sub-domain, emissions were inferred for eight fixed regions, which were not spatially varying. The inversion domain thus spanned 5.2°S–74.1°N, 54.5°E–191.8°E) but results are only presented for the inner sub-domain. Outside of this inversion domain, boundary conditions were estimated as adjustments to vertically uniform 'curtains' on each edge of the inversion domain. Prior values for these curtains were mole fraction baseline data for CHCl₃, calculated using the AGAGE 12-box model from 1997–2016 within latitude bands of 90°N–30°N, 30°N–0°N, 0°S–30°S and 30°S–90°S^{2, 3}. The mean value was used across each latitude band. Though no measurement site within the East Asian domain of this study was used for the model, the stations which were used are positioned to allow good coverage within each semi-hemisphere. By tracking the location of particles exiting the inversion domain, boundary conditions were mapped to these curtains. Adjustments to the boundary conditions were made in the inversion by solving for a scaling to each of these curtains simultaneously with other parameters. CHCl₃ measurement data at HAT and GSN stations were averaged over 24 hours and inversions were run for one year at a time. Prior to averaging, the data were filtered to reduce the contribution of local influence (i.e. unresolved emissions). Using the NAME sensitivity maps, for each measurement, a ratio was determined between the sensitivity local to the station (within 0.47° latitude 0.70° longitude; equivalent to two grid cells) and the total across the

domain. If this local ratio was greater than 10%, the associated measurement point at that time was removed, implying that air was likely stagnant and more prone to local unresolved sources. While fluxes and hyper-parameters (model-measurement uncertainties, prior uncertainties, correlation timescales) were estimated with annual resolution, boundary conditions were estimated on a monthly basis to allow for seasonal variation. Lognormal probability density functions (PDF) used to describe fluxes and boundary conditions and uniform PDFs were used for hyper-parameters. For each inversion, the parameters and hyper-parameters were sampled 100,000 times with an additional 100,000 burn-in iterations, storing every 100th sample for analysis. The simulated CHCl_3 mole fractions are shown alongside observed values for these inversions in Figure S2.

CHCl_3 historical and future demand in China

In China, CHCl_3 is widely used for producing polytetrafluoroethylene (PTFE). The chemical reaction equations for producing PTFE using CHCl_3 are shown below. The capacity of PTFE production increased at 11%/yr between 2010 and 2015, and was projected to increase at 5.3%/yr between 2015 and 2020¹⁴. The PTFE production increased at 17%/yr between 2010 and 2015, and was projected to increase at 7%/yr between 2015 and 2020 (almost a linear increase between 2010 and 2020)¹⁴. Industrial reports of CHCl_3 annual production are currently unavailable. Therefore, PTFE production data in the past (2010–2015) and projected 2015–2020 periods indicates an increasing demand for and production of CHCl_3 in these two periods.





References

1. O'Doherty, S., *et al.* In situ chloroform measurements at Advanced Global Atmospheric Gases Experiment atmospheric research stations from 1994 to 1998. *J. Geophys. Res. Atmos.* **106**, 20429-20444 (2001).
2. Cunnold, D. M., *et al.* The Atmospheric Lifetime Experiment: 3. Lifetime methodology and application to three years of CFC13 data. *J. Geophys. Res. Oceans* **88**, 8379-8400 (1983).
3. Rigby, M., *et al.* Re-evaluation of the lifetimes of the major CFCs and CH₃CCl₃ using atmospheric trends. *Atmos. Chem. Phys.* **13**, 2691-2702 (2013).
4. Spivakovsky, C. M., *et al.* Three-dimensional climatological distribution of tropospheric OH: Update and evaluation. *J. Geophys. Res. Atmos.* **105**, 8931-8980 (2000).
5. Burkholder, J. B., Sander, S. P., Abbatt, J. P. D., Barker, J. R., Huie, R. E., Kolb, C. E., Kurylo, M. J., Orkin, V. L., Wilmoth, D. M. and Wine, P. H. *Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation No. 18* (2015).
6. Rigby, M., Manning, A. J., Prinn, R. G. Inversion of long-lived trace gas emissions using combined Eulerian and Lagrangian chemical transport models. *Atmos. Chem. Phys.* **11**, 9887-9898 (2011).
7. Xiao, X. Optimal Estimation of the Surface Fluxes of Chloromethanes Using a 3-D Global Atmospheric Chemical Transport Model. Ph.D. thesis, Massachusetts Institute of Technology, 2008.
8. Rigby, M., *et al.* Recent and future trends in synthetic greenhouse gas radiative forcing. *Geophys. Res. Lett.* **41**, 2623-2630 (2014).
9. Li, S., *et al.* Emissions of Halogenated Compounds in East Asia Determined from Measurements at Jeju Island, Korea. *Environ. Sci. Technol.* **45**, 5668-5675 (2011).
10. Stohl, A., *et al.* An analytical inversion method for determining regional and global emissions of greenhouse gases: Sensitivity studies and application to halocarbons. *Atmos. Chem. Phys.* **9**, 1597-1620 (2009).
11. Kaminski, T., Rayner, P. J., Heimann, M., Enting, I. G. On aggregation errors in atmospheric transport inversions. *J. Geophys. Res. Atmos.* **106**, 4703-4715 (2001).

- 648
649 12. Thompson, R. L., Stohl, A. FLEXINVERT: an atmospheric Bayesian inversion framework for
650 determining surface fluxes of trace species using an optimized grid. *Geosci. Model Dev.* **7**, 2223-
651 2242 (2014).
- 652
653 13. Trampert, J., Snieder, R. Model Estimations Biased by Truncated Expansions: Possible Artifacts in
654 Seismic Tomography. *Science* **271**, 1257-1260 (1996).
- 655
656 14. Qianzhan. China's fluoropolymer production capacity will be 230,000 tons by 2020, and PTFE will
657 account for 70% (in Chinese). 2017. Available from:
658 <https://www.qianzhan.com/analyst/detail/220/170629-c33a2ca7.html>
- 659
660 15. Chinairn. Price of feedstock for CHCl₃ goes up and demand for CHCl₃ grows (in Chinese). 2014.
661 Available from: <http://www.chinairn.com/news/20140512/164033296.shtml>
- 662
663 16. Yan, C., *et al.* Analysis on market prospects of chlorinated methanes in China (in Chinese with
664 English abstract). *Chlor-Alkali Industry* **45**, 1-4 (2009).

Tables

Table S1. The distance (pmol mol^{-1}) of 90th percentile and 10th percentile of CHCl_3 mole fractions at each station (an approximation of the pollution magnitude) and the corresponding change rates ($\text{pmol mol}^{-1} \text{yr}^{-1}$; using the “least squares” method; the range represents the regression coefficients \pm standard error) over 2007–2015. The change rates for MHD, THD, RPB, SMO and CGO stations are very close to zero, which suggests regional CHCl_3 emissions in Australia (based on CGO station), North America (THD station) and Europe (MHD station) likely did not increase. Change rates for GSN and HAT stations in East Asia are tens and hundreds times larger, which suggests rapid changes in CHCl_3 emissions in East Asia. Subsequently, FLEXPART-based and NAME-based inversions are used to quantify the East Asian CHCl_3 emissions.

	MHD	THD	RPB	SMO	CGO	GSN	HAT
2007	9.6	4.6	2.3	1.7	8.8	16.8	12.0
2008	8.6	5.8	2.8	1.5	8.5	21.9	13.3
2009	7.3	5.3	2.4	1.5	8.2	19.8	15.5
2010	8.6	4.7	2.0	1.7	7.9	25.1	16.3
2011	7.2	4.9	3.0	1.4	8.2	21.3	15.5
2012	7.9	4.9	2.9	1.5	7.8	18.5	18.7
2013	8.4	6.4	3.4	1.5	7.8	34.7	22.8
2014	10.1	6.2	2.9	1.7	9.4	36.8	27.4
2015	9.3	6.0	3.1	1.7	9.6	56.4	27.9
Change rate over 2007–2015	-0.06–0.21	0.07–0.22	0.07–0.15	-0.01–0.02	-0.01–0.17	2.73–4.80	1.80–2.30

679 **Table S2. CHCl₃ emissions (Gg/yr) from each East Asian country/region and the globe.**

		2007	2008	2009	2010	2011	2012	2013	2014	2015
FLEXPART inversion	Eastern China	32 (26–38)	39 (34–45)	30 (25–36)	39 (34–45)	36 (30–42)	42 (36–49)	48 (42–53)	60 (55–65)	88 (80–95)
	Taiwan	1.0 (0.1–1.9)	0.2 (0.0–0.7)	1.5 (0.7–2.2)	0.3 (0.0–1.0)	0.3 (0.0–0.8)	1.1 (0.4–1.9)	0.3 (0.0–0.9)	0.3 (0.0–0.7)	1.4 (0.6–2.2)
	North Korea	0.4 (0.0–1.3)	0.8 (0.0–1.8)	0.3 (0.0–1.0)	0.9 (0.0–2.0)	0.3 (0.0–0.9)	0.8 (0.0–1.9)	1.1 (0.1–2.1)	0.7 (0.0–1.6)	0.9 (0.0–1.9)
	South Korea	1.3 (0.3–2.3)	2.2 (1.4–3.0)	2.1 (1.3–3.0)	0.9 (0.2–1.7)	1.9 (1.1–2.8)	1.4 (0.6–2.2)	1.1 (0.4–1.8)	0.7 (0.1–1.4)	2.9 (2.0–3.9)
	Japan	4.7 (1.8–7.6)	4.4 (1.8–7.1)	4.7 (2.3–7.1)	3.6 (0.6–6.5)	3.6 (1.1–6.1)	2.3 (0.1–4.4)	4.0 (1.6–6.4)	6.1 (3.7–8.5)	2.2 (0.2–4.2)
NAME inversion	Eastern China	51 (45–57)	49 (43–54)	40 (35–44)	32 (26–40)	55 (49–62)	51 (38–64)	63 (56–69)	81 (72–90)	82 (70–101)
	Taiwan	1.2 (0.4–2.2)	0.3 (0.0–0.6)	0.6 (0.2–1.2)	0.2 (0.0–0.3)	0.3 (0.1–0.8)	0.3 (0.0–0.6)	0.4 (0.0–1.0)	0.6 (0.0–1.6)	0.7 (0.2–1.6)
	North Korea	0.9 (0.2–2.1)	0.9 (0.4–1.7)	0.9 (0.3–1.8)	0.3 (0.0–0.9)	1.5 (0.6–2.6)	1.0 (0.1–2.2)	1.1 (0.4–2.4)	1.6 (0.7–2.6)	2.6 (1.1–4.3)
	South Korea	1.0 (0.3–1.7)	1.4 (0.8–2.1)	2.0 (1.5–2.6)	2.1 (1.7–2.5)	1.8 (1.4–2.4)	1.8 (1.3–2.3)	1.2 (0.4–2.2)	2.0 (1.4–2.7)	2.1 (0.9–3.7)
	Japan	5.1 (1.7–8.5)	5.2 (3.0–7.9)	4.0 (1.8–6.0)	1.9 (0.4–4.6)	11.8 (7.7–16.6)	2.8 (0.7–4.8)	1.3 (0.2–2.9)	8.6 (5.0–12.4)	7.5 (3.7–10.9)
AGAGE 12- box inversion	Globe	277 (222–340)	272 (219–332)	259 (209–320)	279 (227–341)	279 (222–340)	290 (234–356)	306 (247–373)	321 (256–392)	324 (261–397)

681 **Table S3. Location information of major CHCl₃ factories in China^{15, 16} (may not be complete;**
682 **latitude/longitude information was obtained by using google earth).**

Factory number	Latitude	Longitude
1	37.0	118.5
2	31.7	121.0
3	29.3	104.8
4	37.1	119.0
5	37.0	118.0
6	32.5	119.9
7	36.4	116.2
8	28.9	118.9
9	32.2	119.6

683

684 **Table S4. Information of *in situ* CHCl₃ measurement sites used in this study.**

Station	Code	Latitude	Longitude	Altitude (m a.s.l.)	Calibration scale
Mace Head, Ireland	MHD	53.3°N	9.9°W	5	SIO-98
Trinidad Head, California, USA	THD	41.1°N	124.2°W	107	SIO-98
Ragged Point, Barbados	RPB	13.2°N	59.4°W	45	SIO-98
Cape Matatula, American Samoa	SMO	14.2°S	170.6°W	77	SIO-98
Cape Grim, Tasmania, Australia	CGO	40.7°S	144.7°E	94	SIO-98
Gosan, South Korea	GSN	33.3°N	126.2°E	72	SIO-98
Hateruma, Japan	HAT	24.1°N	123.8°E	47	NIES-11

685

Table S5. Performance of FLEXPART and NAME inversions on simulating CHCl₃ mole fractions at HAT and GSN stations. *B* represents the mean bias (pmol mol⁻¹) between the simulations and measurements (simulated values minus measurements). *r* represents Pearson correlation coefficients between the simulations and measurements. *RMSE* represents the root mean square error (pmol mol⁻¹) between the simulations and measurements.

Station	Year	<i>B</i>		<i>r</i>		<i>RMSE</i>	
		FLEXPART inversion	NAME inversion	FLEXPART inversion	NAME inversion	FLEXPART inversion	NAME inversion
HAT	2007	-0.73	1.37	0.80	0.81	3.42	3.81
	2008	-0.52	0.91	0.84	0.85	3.07	3.35
	2009	-0.84	-0.07	0.80	0.89	3.28	3.01
	2010	-0.29	0.66	0.89	0.84	3.54	4.31
	2011	-0.99	0.55	0.84	0.91	3.98	3.09
	2012	-0.93	1.35	0.90	0.89	3.88	4.35
	2013	-2.13	0.06	0.90	0.89	5.21	4.95
	2014	-0.80	0.46	0.93	0.88	4.05	5.58
	2015	-0.14	0.89	0.92	0.84	4.12	6.36
GSN	2007	-0.90	-1.32	0.68	0.84	4.94	4.25
	2008	-0.87	-1.58	0.83	0.82	5.17	5.64
	2009	-2.02	-1.10	0.69	0.83	6.77	6.17
	2010	-2.08	-1.62	0.89	0.82	7.31	6.49
	2011	-1.11	1.00	0.75	0.63	5.17	8.37
	2012	-1.79	0.50	0.78	0.85	5.99	5.03
	2013	-3.93	-3.86	0.79	0.75	11.40	10.73
	2014	-3.06	-1.00	0.79	0.85	10.64	8.20
	2015	-6.22	-8.34	0.80	0.73	17.41	22.83

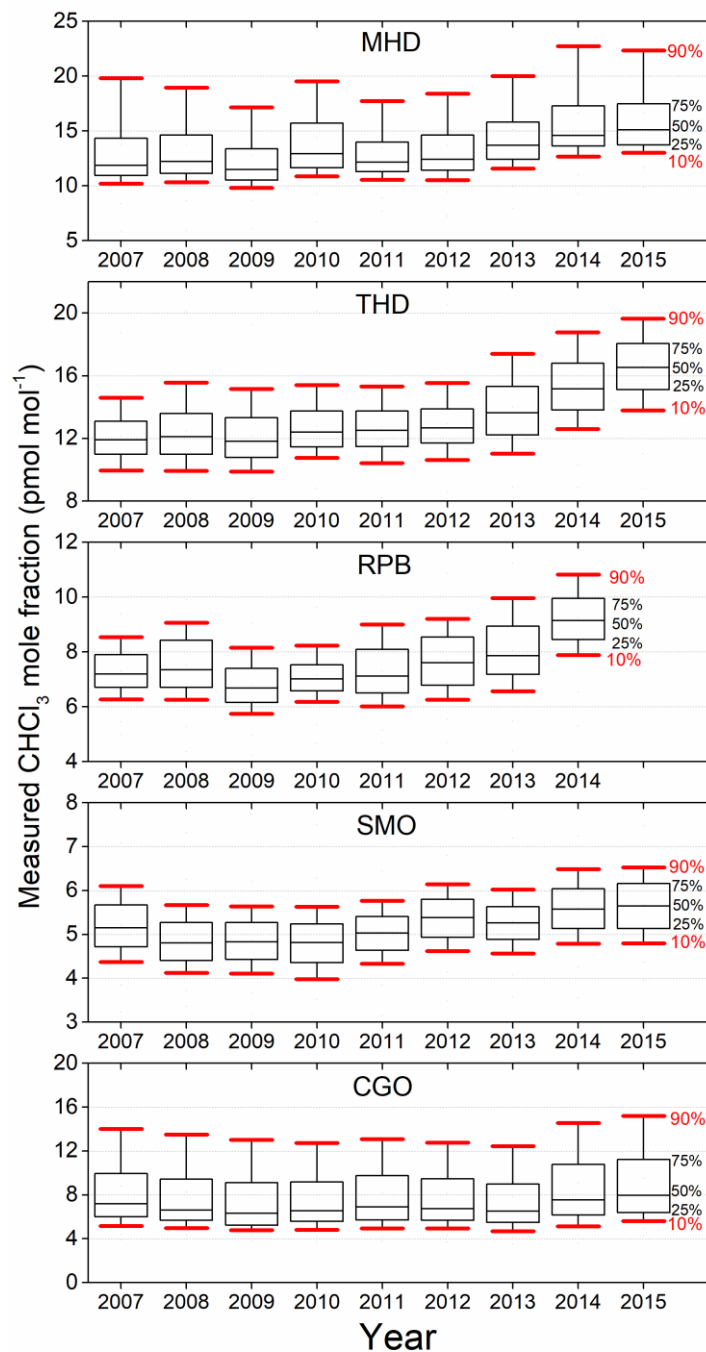
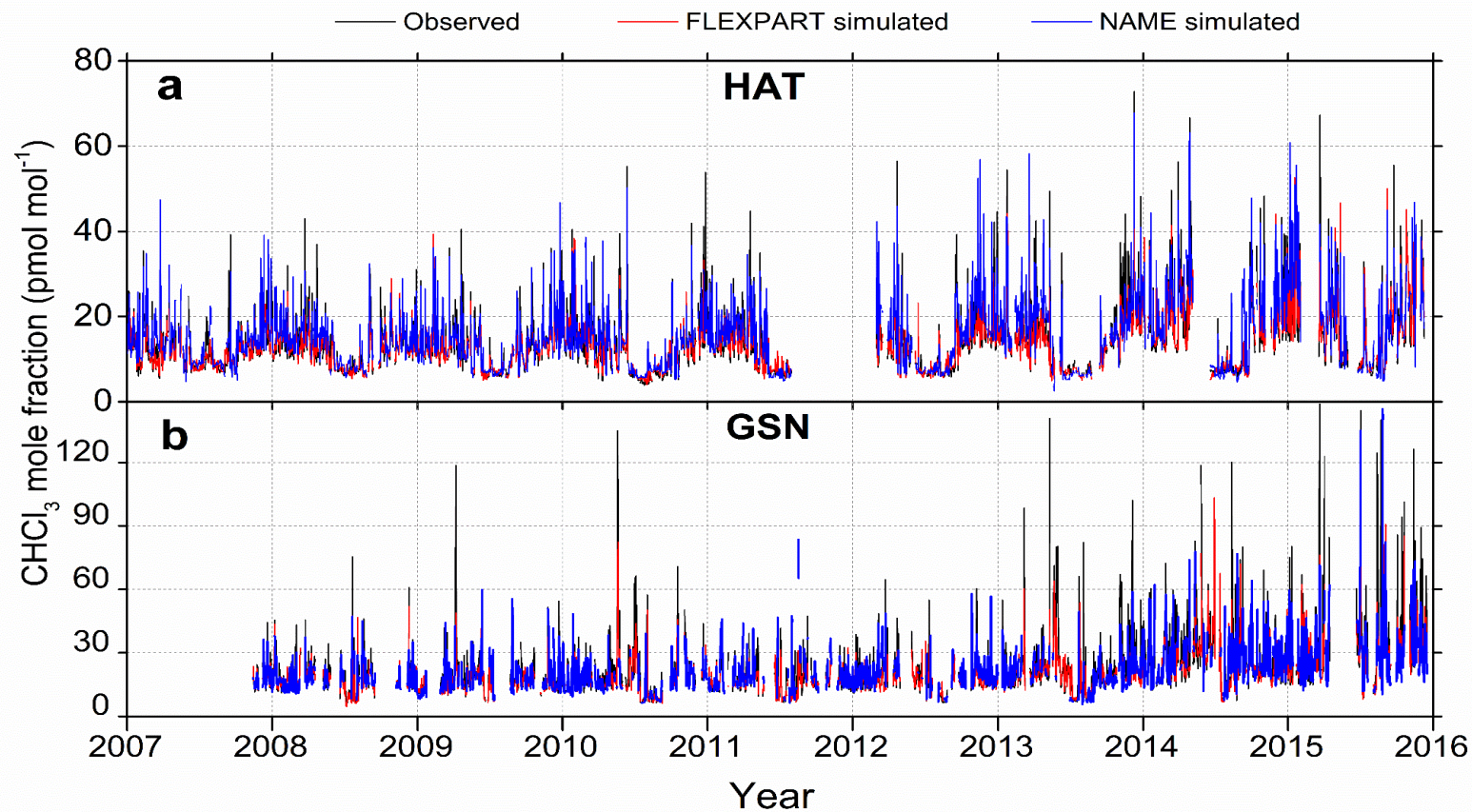


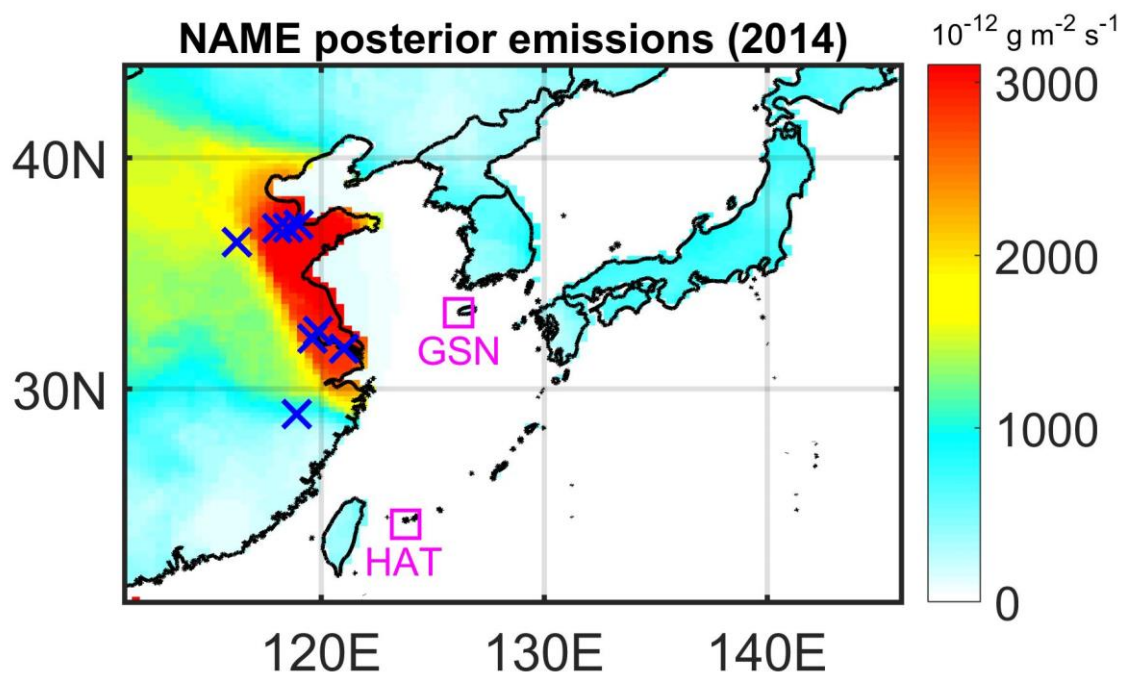
Figure S1. Box and whisker plots for the measured CHCl_3 mole fractions at MHD, THD, RPB, SMO and CGO stations. The 10th, 25th, 50th, 75th and 90th percentiles are shown. The distance of 90th percentile and 10th percentile is an approximation of the pollution magnitude and is related to the regional emission strength of CHCl_3 . RPB data for 2015 are not included due to biased low CHCl_3 mole fraction values induced by no measurements in some months in 2015. Statistics of enhanced CHCl_3 mole fractions for each station, as well as GSN and HAT stations, are provided in Table S1.



700

701 **Figure S2. Observed and simulated CHCl₃ mole fractions at HAT and GSN stations from FLEXPART and NAME inversions.** Statistics of
 702 performance of FLEXPART and NAME inversions on simulating CHCl₃ mole fractions for each station are provided in Table S5. Note that some
 703 measurement points were excluded by a filter algorithm in NAME inversions (see main text).

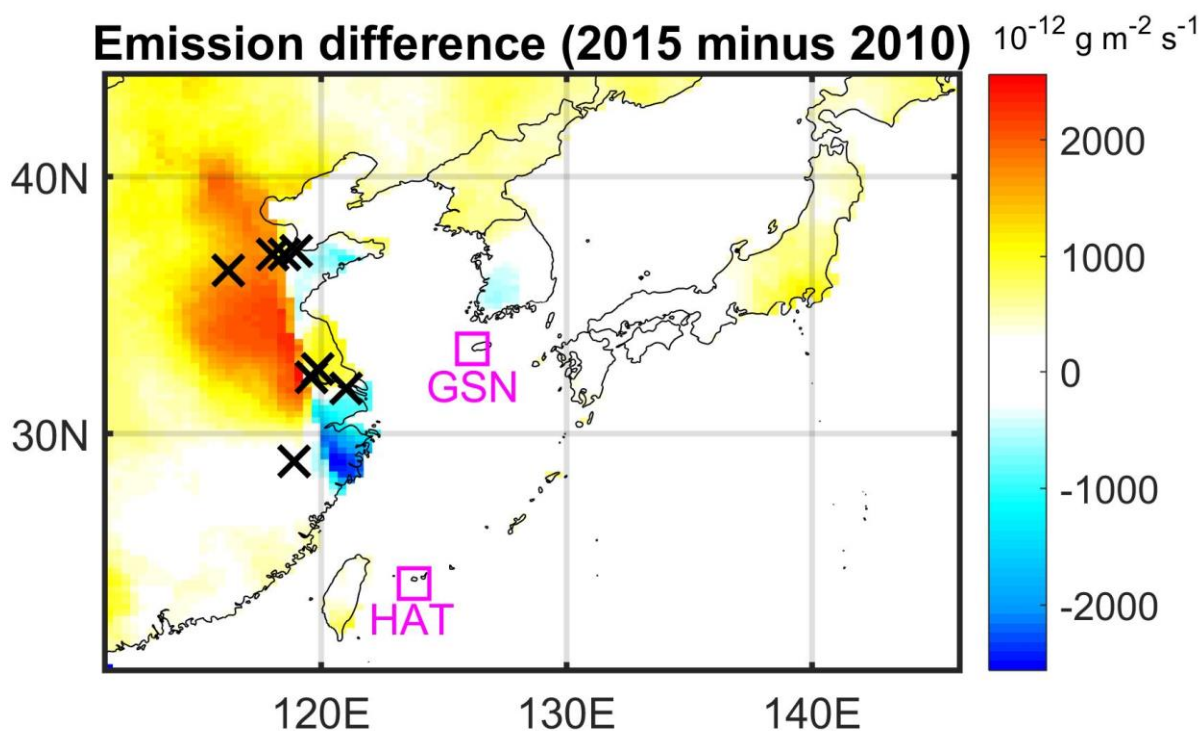
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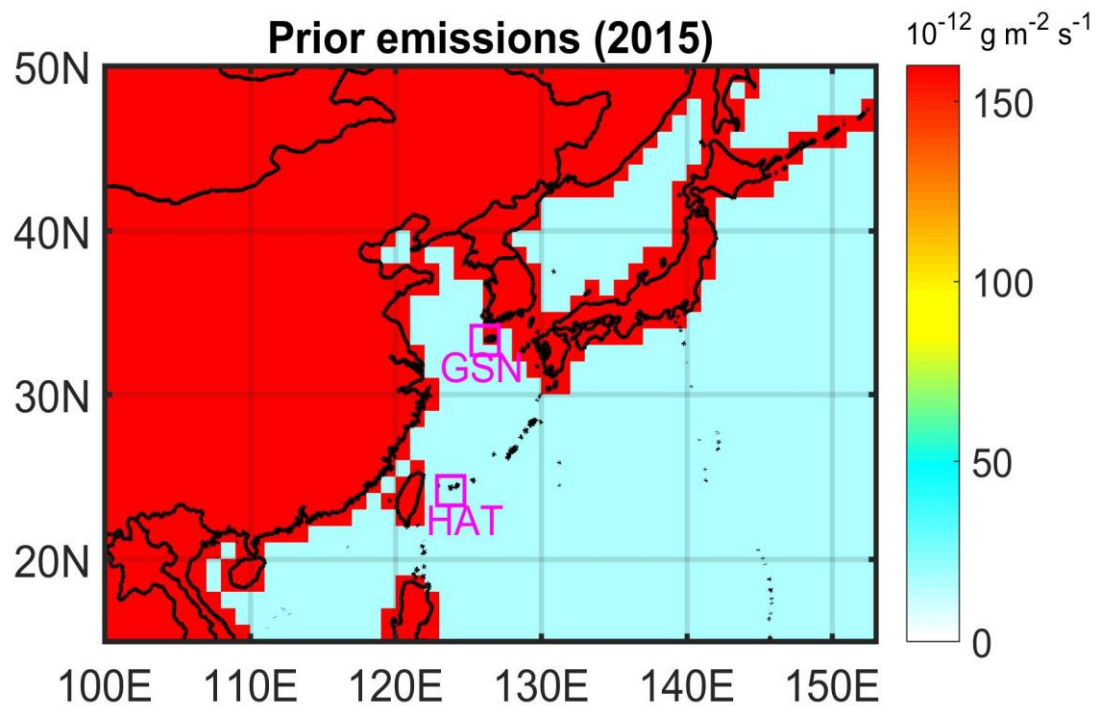
706 **Figure S3. Map of posterior CHCl_3 emissions derived from NAME inversion.** The blue crosses
 707 represent the factories of CHCl_3 production and the purple squares represent the measurement stations.

708



709

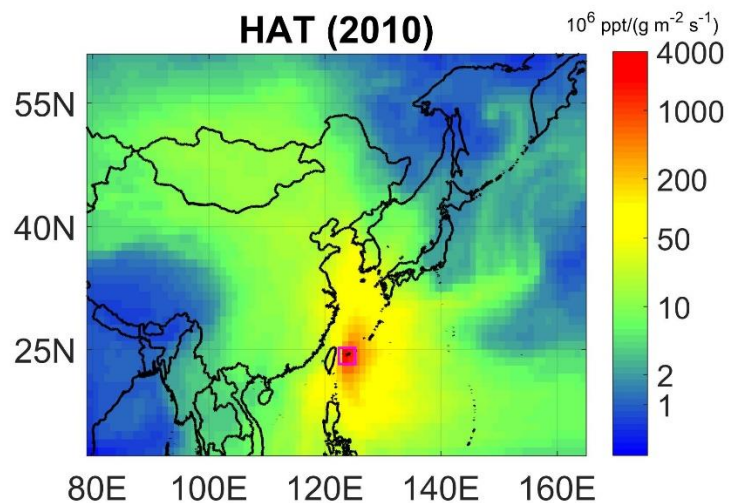
710 **Figure S4. Spatial differences of posterior emissions between 2015 and 2010 (2015 minus 2010;**
 711 **NAME-based inversions).** The black crosses represent the factories of CHCl_3 production and the purple
 712 squares represent the measurement stations.



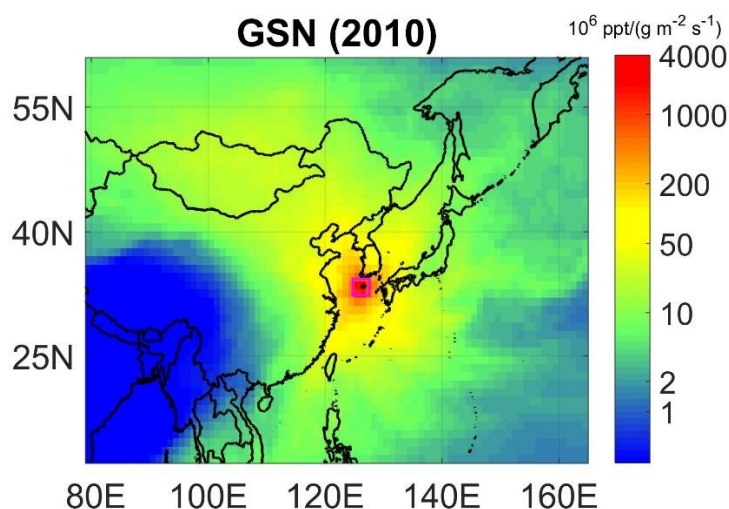
713

714 **Figure S5. Map of prior CHCl_3 emissions from continent and ocean used in FLEXPART and**

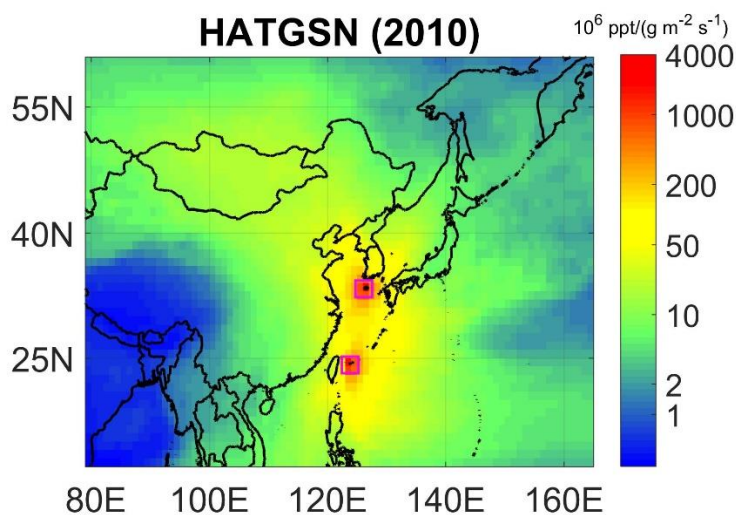
715 **NAME inversions.** Prior emissions are the same for all years during 2007–2015.



716



717



718

719 **Figure S6. Annual average emission sensitivity from FLEXPART simulations for HAT, GSN**
 720 **and HATGSN stations for the year 2010. The purple squares represent the measurement stations.**

